

## Calcium fluoride ( $\text{CaF}_2$ )



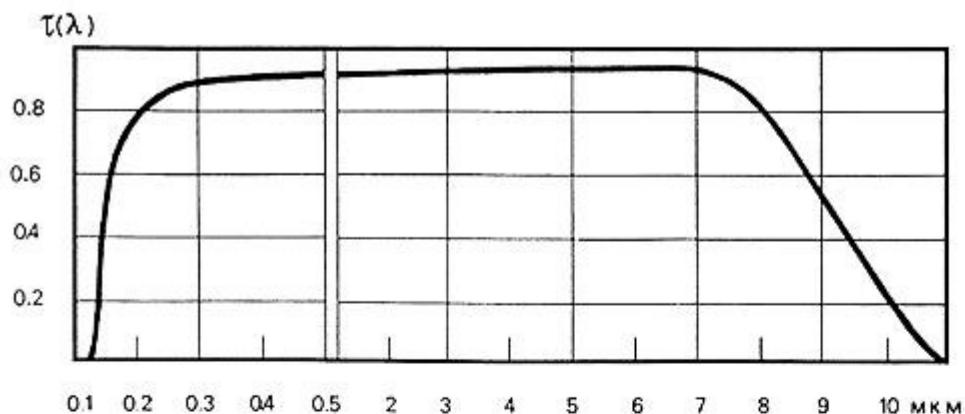
### 1. Crystallographic characteristics

Crystal system group	Class	Lattice parameters		Cleavage
		a	c	
Cubic	$m\bar{3}m$	5.462	a	Perfect along (111)

### 2. Optical characteristics

$n_e$	$n_F - n_C$	$n_{10,5}$	$n_{8,0} - n_{12,5}$	Internal transmittance coefficient		Refractive index	
				$l$ , mkm	$t_i$ (1)	$l$ , mkm	$n$ (1)
1.4349	0.0043	1.2996	-				
Attenuation coefficient $\alpha$ (1), $\text{cm}^{-1}$ , less than						0.2	1.4951
						0.5	1.4365
CF - U	CF - V	CF - I		0.2	0.87	1.0	1.4289
0.10 (0.2 mkm)	0.01 (0.4 mkm)	0.05 (7.5 mkm)		0.5	0.97	2.0	1.4239
Thermal coefficient of refractive index at $\alpha = 3.39$ mkm, $10^{-5} \text{ }^{\circ}\text{C}^{-1}$ in range of $\pm 60^{\circ}\text{C}$				1.0	0.99	3.0	1.4179
				3.0	0.99	4.0	1.4098
				5.0	0.99	5.0	1.3990
				6.0	0.98	6.0	1.3856
				7.0	0.87	7.0	1.3693
				8.0	0.88	8.0	1.3498
				9.0	0.59	9.0	1.3268
				10.0	0.19	10.0	1.3002
						11.0	1.2676
						12.0	1.2299
(-0.95) - (-1.17)		0.15 - 9.0					

## Transmission spectrum



## 3. Thermal characteristics

Thermal expansion $\alpha_t$ , $10^8 \cdot ^\circ\text{C}^{-1}$	Thermal conductivity, $\text{W}/(\text{m} \cdot ^\circ\text{C})$	Specific heat, $10^3 \text{ J}/(\text{kg} \cdot ^\circ\text{C})$	Heat resistance, $^\circ\text{C}$	Melting Point, $^\circ\text{C}$
in range of $\pm 60^\circ\text{C}$	at $38^\circ\text{C}$	at $40^\circ\text{C}$	$20 \pm 2$	1418
$16.2 \div 19.4$	9.71	0.8876		

## 4. Mechanical characteristics

Density, at $20^\circ\text{C}$ , $\text{g}/\text{cm}^3$	Hardness/Mohs	Microhardness, $10^7 \text{ Pa}$	Elastic compliance constants, $10^{-12} \text{ Pa}^{-1}$		
			$S_{11}$	$S_{12}$	$S_{44}$
3.18	4	165	6.83	-1.53	29.58
Modulus of elasticity $E$ , in direction, $10^{10} \text{ Pa}$ ,		Shear modulus $G$ , in plane, $10^{10} \text{ Pa}$ ,	Transverse deformation coefficient $n$		
<100>	<111>	(100)	(111)		
14.61	8.99	4.76	3.38	0.216	

Optic stresses coefficient		Photoelastic constants		Piezooptic constants at $n = 0.589 \text{ mkm}, 10^{-12} \text{ Pa}^{-1}$		
$n = 0.589 \text{ mkm}, 10^{-12} \text{ Pa}^{-1}$						
B <sub>1</sub>	B <sub>2</sub>	C <sub>1</sub>	C <sub>2</sub>	n <sub>11</sub>	n <sub>12</sub>	n <sub>44</sub>
2.14	-1.03	0.43	-1.71	-0.29	1.16	0.70

## 5. Chemical stability

Solubility		
in water at 20°C, g/100 cm <sup>3</sup>	in acids	in organic solvents
0.0016	low soluble	not soluble in acetone